Name : _____ Date : _____

Atomic Structure & Periodicity

| [1] | What did JJ Thomson discover? | | | | | |
|------|---|-------------------------------|-------------------------------|---------------------------------|--|--|
| | a) Proton | b) Neutron | c) Quark | d) Electron | | |
| [2] | Millikan discovered the mass and charge of which particle? | | | | | |
| | a) Proton | b) Neutron | c) Electron | d) Alpha particle | | |
| [3] | Which of the following types of radiation has the longest wavelength? | | | | | |
| | a) gamma rays | b) visible | c) ultraviolet | ^{d)} radar | | |
| [4] | Which of the following has the longest wavelength? | | | | | |
| | a) blue light | b) yellow light | c) red light | ^{d)} green light | | |
| [5] | Which of the following types of radiation has the highest energy? | | | | | |
| | a) gamma rays | b) visible | c) ultraviolet | ^{d)} radar | | |
| [6] | Which of the following is NOT a proper unit of frequency? | | | | | |
| | a) Hz | b) s ⁻¹ | c) m·s -1 | d) 1/s | | |
| [7] | What is the relationship between the energy of a photon of light and its frequency? | | | | | |
| | a) E = ν | b) $E = h/v$ | c) $E = h\nu$ | d) $E = 1/h\nu$ | | |
| [8] | What is the de E speed of light? | noving at 80% of the | | | | |
| | a) $3.03 \times 10^{-12} \text{ m}$ | b) 2.42 x 10 ⁻¹² n | n c) 3.30 x 10 ⁻¹¹ | m d) 1.59 x 10 ⁻²⁵ m | | |
| [9] | Which quantum number determines the subshell occupied by the electron | | | | | |
| | a) n | b) | c) m ₁ | d) \mathbf{m}_{s} | | |
| [10] | Which of the following electronic transitions in a hydrogen atom would have the longest wavelength? | | | | | |
| | a) $n = 4$ to $n = 1$ | b) n = 4 to n = 2 | c) n = 2 to n = | = 1 d) n = 4 to n = 3 | | |

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| Answers | | | | | | | | |
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| | a) $E = v$ | b) $E = h/v$ | c)E = hv | d) $E = 1/h\nu$ | | | | |
| [8] | What is the de Broglie wavelength of an electron moving at 80% of the speed of light? | | | | | | | |
| | (a) 3.03 x 10 ⁻¹² m | b) 2.42 x 10 ⁻¹² | m c) 3.30 x 10 ⁻¹ | ¹ m ^{d)} 1.59 x 10 ⁻²⁵ m | | | | |
| [9] Which quantum number determines the subshell occupied by the electron? | | | | | | | | |
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