Basic Atomic Structure Practice Worksheet

1.	What is the atomic mass of Carbon:			
2.	What is the atomic symbol for Sodium:			
3.	What is the number of neutrons in Lithium:			
4.	What is the number of electrons in Neon:			
5.	What is the number of protons in Boron:			
6.	What is the atomic mass of Helium:			
7.	Give the symbol and number of protons in one atom of:			
	Carbon	Neon	_	Calcium
	Gallium	Silver		Zinc
	Platinum	Radium		
8.	Give the symbol and number of neutrons in one atom of:			
	Fluorine – 19	Zi	rconium – 91 _	
	Aluminum – 27	Ar	ntimony - 121 _	
	Argon – 40	Tu	ıngsten – 184 _	
	Manganese - 55	Ra	ndium - 226	
9.	Write the complete chemical symbol for the ion with 75 protons, 111 neutrons, and 71 electrons?			
10.	^{10.} Write the complete chemical symbol for the ion with 26 protons, 29 neutrons, and 24 electrons?			

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Answers

- 1. What is the atomic mass of Carbon: 12.001
- 2. What is the atomic symbol for Sodium: Na
- 3. What is the number of neutrons in Lithium: 4
- 4. What is the number of electrons in Neon: 10
- 5. What is the number of protons in Boron: ____5___
- 6. What is the atomic mass of Helium: 4.003
- 7. Give the symbol and number of protons in one atom of:

Carbon <u>C 6</u>

Neon <u>Ne 10</u>

Calcium Ca 20

Gallium Ga 31

Silver Ag 47

Zinc Zn 30

Platinum Pt 78

Radium Ra 88

8. Give the symbol and number of neutrons in one atom of:

Fluorine – 19 F 10

Zirconium – 91 Zr 51

Aluminum – 27 Al 14

Antimony - 121 <u>Sb 70</u>

Argon – 40 Ar 22

Tungsten – 184 W 110

Manganese - 55 Mn 30

Radium - 226 <u>Ra 138</u>

- 9. Write the complete chemical symbol for the ion with 75 protons, 111 neutrons, and 71 electrons?
 186Re⁴⁺
- $^{10.}$ Write the complete chemical symbol for the ion with 26 protons, 29 neutrons, and 24 electrons? 55 Fe²⁺